

### Formation: Sources of Evidence



#### Star-forming regions • Chemistry of

source material



#### Our solar system

- 1. Patterns of motions
- 2. 2 types of planets
- 3. Asteroids and comets
- 4. Exceptions



Other solar systems •Similarities and differences



http://exoplanets.org/table/

## Collapse of the Solar Nebula

- Formation of the Sun seems a good place to start.
- Theories of star formation are based on observing millions of stars of different ages.
- Start with a nebula of gas and dust.
  - Nebula = noun = "cloud" (plural = nebulae).
  - Nebular = adjective = "cloud-like"

Section could have been called Collapse of Nebular Solar Nebula.

# How Big Was Solar Nebula?

- ~1% efficiency (guess)
- start with ~100 Mass of Sun = 10<sup>32</sup> kg
- If Temperature of cloud ~1000K
  - density ~ 10<sup>-12</sup> kg m<sup>-3</sup>
  - R~2,500 AU
- If Temperature of cloud ~10K
  - density ~  $10^{-18}$  kg m<sup>-3</sup>
  - R~250,000 AU

Element	Symbol	Atomic #	Relative Abundance	
Hydrogen	Н	1	1,000,000	
Helium	He	2	80,000	
Carbon	С	6	420	
Nitrogen	N	7	87	
Oxygen	0	8	690	
Neon	Ne	10	130	
Sodium	Na	11	2	
Magnesium	Mg	12	32	
Aluminum	Al	13	3	
Silicon	Si	14	45	
Sulfur	S	16	16	
Argon	Ar	18	1	
Calcium	Ca	20	2	
Iron	Fe	26	32	
Nickel	Ni	28	2	

#### Cosmic Abundances of Elements

What was the solar nebular made of?

# Meteorites vs. Solar Photosphere

- Note the striking similarity between meteoritic and photospheric compositions
- Normally gaseous elements (N,C,O) are enriched in photosphere relative to meteorites
- We can use this information to obtain a bestguess nebular composition



Basaltic Volcanism Terrestrial Planets, 1981









## **Galactic Recycling**



- •Many fusion products remain locked up in "stellar corpses"
- •Stellar winds and explosions are the main ways fusion products escape
- •Escaping gases mix in giant molecular clouds, enriching them in heavy elements ("metals")
- •More stars (and solar systems?) are born...
- •...and die, continuing the recycling and enrichment

## **Nebular Composition**

 Based on solar photosphere and chondrite compositions, we can come up with a best-guess at the nebular composition (here relative to 10<sup>6</sup> Si atoms):

Element	Н	He	С	Ν	0	Ne	Mg	Si	S	Ar	Fe
Log <sub>10</sub> (No. Atoms)	10.44	9.44	7.00	6.42	7.32	6.52	6.0	6.0	5.65	5.05	5.95
Condens. Temp (K)	180		78	120			1340	1529	674	40	1337

- Blue are volatile (easy to vaporize), red are refractory (hard to vaporize)
- Most important refractory elements are Mg, Si, Fe, S

Data from Lodders and Fegley, *Planetary Scientist's Companion*, CUP, 1998 This is for all elements with relative abundances  $> 10^5$  atoms.

		ADLE 2.0.1. Interstenar in	in market in the second s	
Simple Hydrides, Ox	vides, Sulfides, Halides, an	d Related Molecules		
H <sub>2</sub>	co	NH <sub>3</sub>	CS_	NaCl
HČI	SiO	SiHi	585	AICI
H <sub>2</sub> O	SO2	CC	H <sub>2</sub> S	KCI
-	OCS	CH₄	PN	AlF
Nitriles, Acetylene L	Derivatives, and Related M	tolecules		
HCN	IIC=C-CN	$H_3C - C = C - CN$	H <sub>3</sub> CCH <sub>2</sub> CN	H <sub>2</sub> C=CH <sub>2</sub>
H <sub>1</sub> CCN	$H(C=C)_2 - CN$	H <sub>3</sub> C-C=CH	H <sub>2</sub> C=CH-CN	HC=CH
céco	$H(C=C)_{s}-CN$	H <sub>4</sub> C−(C≡C) <sub>2</sub> −H	HNC	
OCCS	$H(C=C)_4 - CN$	3 I I I I I I I I I I I I I I I I I I I	HN=C=O	
HC=CCHO	H(C=C)_CN		HN=C=S	
H <sub>3</sub> CNC				
Aldebydes, Alcobols	, Ethers, Ketones, Amides,	and Related Molecules		
H <sub>2</sub> C=O	H <sub>4</sub> COH	HO-CH=O	H <sub>2</sub> CNH	
H <sub>2</sub> C=S	H <sub>4</sub> C-CH <sub>2</sub> -OH	H <sub>4</sub> C-O-CH=O	H <sub>3</sub> CNH <sub>2</sub>	
H <sub>*</sub> C-CH=O	H <sub>a</sub> CSH	H <sub>2</sub> C-O-CH <sub>3</sub>	H <sub>2</sub> NCN	
NH2-CH=O	(CH <sub>3</sub> )2CO?	H <sub>z</sub> C=C=O	-	
Cyclic Molecules				
C <sub>3</sub> H <sub>2</sub>	SiC <sub>2</sub>	C <sub>3</sub> H		
lons				
CH+	HCO+	HCNH+		
HN <sub>2</sub> +	HOCO+ HCS+	SO+		
Radicals				
OH	C <sub>3</sub> H	CN	HCO	C <sub>2</sub> S
CH	C <sub>4</sub> H	C <sub>4</sub> N	NO	NS
C <sub>2</sub> H	C <sub>4</sub> H	H <sub>2</sub> CCN	SO	
0,11	C <sub>6</sub> H			

TABLE 2.6.1. Interstellar molecules.

Data from Irvine W. M. and Knacke R. E. (1989) in Origin and Exclusion of Planetary and Satellite Atmospheres (S. Atreya p. 5, Fig. 1, Univ. of Arizona, Tueson.

### Images of Protoplanetary Disks



a We see this disk edge-on around the star AU Microscopii, confirming its flattened shape.

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b This photo shows a disk around the star HD141569A. The colors are not real; a black-and-white image has been tinted red to bring out faint detail.

We can now watch solar systems being formed...

### **Oldest Meteorite**



Allende - fell to Earth near Chihuahua, Mexico at 1:05am on February 8, 1969. Age: 4.5 BY old

How do we know it's that old?





The best age for the Earth (4.54 Ga) is based on Pb (lead) ratios in troilite from iron meteorites, specifically the Canyon Diablo meteorite.

Mineral grains (zircon) with U-Pb ages of 4.4 Ga have recently been reported from sedimentary rocks in west-central Australia.

# Oldest Rocks



### Oldest Rocks

- The oldest dated lunar rocks have ages between 4.4 and 4.5 billion years and provide a minimum age for the formation of Moon
- The meteorites, and therefore the Solar System, formed between 4.53 and 4.58 billion years ago
- <u>http://pubs.usgs.gov/gip/geotime/</u> <u>radiometric.html#table</u>





Spin up: Conservation of Angular Momentum MVR = Constant



Where did the angular momentum come from???

Small random motions averaging out to a tiny bulk motion

- this bulk motion is then "amplified" (due to conservation of angular momentum) as the cloud collapses





Sun lost most angular momentum -Most left in *orbital* momentum of Jupiter (& Saturn)

Figure 111.7 Orbital angular momenta of the planets. Note the overwhelming importance of the Jovian planets.



# **Cartoon of Nebular Processes**



- · Scale height increases radially
- Temperatures decrease radially consequence of lower irradiation, and lower surface density and optical depth leading to more efficient cooling





### **Temperature and Condensation**

Nebular conditions can be used to predict what components of the solar nebula will be present as gases or solids:





Temperature profiles in a young (T Tauri) stellar nebula, D'Alessio et al., *A.J.* 1998

Condensation behaviour of most abundant elements of solar nebula e.g. C is stable as CO above 1000K,  $CH_4$  above 60K, and then condenses to  $CH_4.6H_2O$ . From Lissauer and DePater, *Planetary Sciences* 

### The Cosmic Abundance of Elements

Element	Symbol	Atomic Number	Number of Atoms per Million Hydrogen Atoms	The Astronomer's Periodic Table
Hydrogen	н	1	1,000,000	(Ben McCall)
Helium	He	2	68,000	
Carbon	С	6	420	
Nitrogen	N	7	87	
Oxygen	0	8	690	
Neon	Ne	10	98	
Sodium	Na	11	2	
Magnesium	Mg	12	40	He
Aluminum	AI	13	3	пе
Silicon	hi \	14	38	
Sulfur	s	16	19	
Argon	Ar	18	4	
alcium	Ca	20	2	
ron	Fe	16	34	
Nickel	Ni	28	2	CNO Ne
	ш			. Mg Si S Ar
	H	,U ľ	NH <sub>3</sub> CH₄	

Water, Ammonia, Methane  $CO_2$  CO  $N_2$ 

Ignore inert gases He, Ne, Ar

5

### Minimum Mass Solar Nebula



Fig. 1.1. The surface density in gas (upper line) and solids (lower broken line) as a function of radius in Hayashi's minimum mass Solar Nebula. The dashed vertical line denotes the location of the snow line.



- 98% of material hydrogen & helium does not condense.
- Inside frostline only refractory materials condense rocks & metals
- Outside frostline volatiles also condense WAM AND rocks & metals too.



### Planetesimals

- Oligarchic growth the bigger get bigger
- Beyond the frost line snowballs snowball....
- REALLY big (icy) planetesimals (~20 Rearth) gravitationally pull in hydrogen - the most abundant gas - and become GIANT.



- Lasts ~ 10<sup>6</sup> years or less near 5 AU
- Can produce 10 M<sub>Earth</sub> cores
- Requires disk densities 2-3 times higher than that predicted by minimum mass solar nebula



Stage 2: steady gas infall

 Controlled by balance of radiative cooling and heating from contraction and planetesimal accretion

• Continues as long as planetary envelope is in contact with protosolar nebula



Stage 4: 'Isolation' phase



- Planet continues to contract and cool

# Why Only 2 3 Types of Planets?

- 1. Cosmic Abundance of Elements
- 2. Temperature Colder Farther from Sun
  - Abundance ices condense beyond frost line
  - Snowballs -> bigger snowballs
  - Giant snowballs have enough gravity to hold H most abundant element - > giant planets
  - Small amounts of rock & metal-> terrestrial planets
  - Ice dwarf planets, comets, asteroids = leftovers

## Terrestrial (silicate) planets



- Consist mainly of silicates ((Fe,Mg)SiO<sub>4</sub>) and iron (plus FeS)
- Mercury is iron-rich, perhaps because it lost its mantle during a giant impact (more on this later)
- Volatile elements (H<sub>2</sub>O,CO<sub>2</sub> etc.) uncommon in the inner solar system because of the initially hot nebular conditions
- · Some volatiles may have been supplied later by comets
- Satellites like Ganymede have similar structures but have an ice layer on top (volatiles are more common in the outer nebula)

# **Gas Giant Planets**



Figure from Guillot, *Physics Today*, (2004). Sizes are to scale. Yellow is molecular hydrogen, red is metallic hydrogen, hydrogen compounds are blue, rock is grey.

# But there's still **gas** between planets

What happens to the gas?
Young stars go through phase of v. strong solar wind
blows away gas



# Phases of SS Formation

- Dense cloud collapse (0.1-0.5My)
- Disk dissipation (material falls onto disk & is transported onto Sun) (0.05 My)
- Protosun becomes a T Tauri star; accretion begins (1-2My)
- Nebular gas loss by T Tauri winds (3-30My)

### But there's still junk between planets

What happens to the junk?

- Gets kicked about mostly by JSUN
- Kicked out of solar system
- Herded asteroid belt, Kuiper Belt
- Captured as moons
- Bashed into young planets -> Moon, Charon, others?
- Delivers water (+other volatiles) to Earth



http://www.minorplanetcenter.net/iau/lists/InnerPlot.html

distance from Sun (AU)







# Can migration and planetesimal scattering explain the "exceptions"?

#### Uranus tipped on side

#### Late Heavy Bombardment:

Leftover planetesimals bombarded other objects in the late stages of solar system formation

**Earth oceans:** Water & other volatiles must have come to Earth by way of icy planetesimals from beyond the frost line



### Issues addressed:

- 1. Angular momentum distribution
- 2. Orderly motion
- 3. 3 types of planets
- 4. Time scale of formation
- 5. Asteroids, comets, Kuiper Belt
- 6. How Earth got oceans
- Anomalies? Uranus tipped on side?
   Formation of Moon & Charon?
   Collisions can explain anything!



• Does Earth just have more vigorous convection or is something else involved?

• Could **liquid water** be involved? Earth has lots, Venus has none (lost due to run-away greenhouse, dissociation, ionization, solar wind stripping)

### Are the plates pushed - or do they pull?



### Differences in surface geology – and dynamo due to their atmospheres?

Watery Earth: vigorous recycling of crust / upper mantle
driven by "eclogite engine"

• Dry, hot Venus: static crustal lid holds in heat generated by radioactivity in mantle until erupts in (periodic?) episode(s?) of volcanic resurfacing

Heat flux from core driven by mantle convection:
Earth – strong mantle convection, large temperature gradient, drives core convection -> dynamo
Venus – low temperature gradient, stablely stratified core -> no dynamo

...and WHY WHY WHY

is Earth different?

### Requirements for Life\*

- Organic Chemicals
- Energy
- Liquid water
- Moderate stability
- over millions of years
- Nothing too nasty radiation, changes in climate, mass destruction, epidemic diseases....

\* based on extrapolation from one example - Earth!



#### Stability of Earth's Climate thanks to the CO<sub>2</sub> cycle & FEEDBACK MECHANSIMS

- Allowed Earth to stay 'habitable' despite Sun brightening over billions of years
- Takes half a million years to stabilize...



# Why is Earth Habitable?

- Geology: plate tectonics, erosion
- Water: abundant and liquid
- Atmosphere: oxygen, stratosphere, little CO<sub>2</sub>
- Stable climate (unlike Mars & Venus)
- Life: astonishing and planet-altering

# All of these unique features are interrelated

# Origin of Life

- Almost 4 billion years ago early often?
- · Have evidence for early life
- No one knows how
- Experiments show nature can make life's building blocks





## **Earliest Fossils**



- Oldest fossils (bacterialike organisms)
- 3.4 billion yrs ago
- Carbon isotope evidence pushes origin of life to >3.85 billion yrs ago
- Life arose about as soon as possible after bombardment!!!

Fossil algae = stromatolites

# Tree of Life



- "tree of life" from DNA study
- Plants & animals a tiny part of tree.
- Common Ancestor most likely arose at "black smoker" near a volcanic region of tectonic plate spreading



#### 65 million years ago in Colorado & around the world

Most species alive, including dinosaurs, disappeared from earth



#### Sedimentary rock layer from that time shows:

- Iridium & other elements SO WHAT?? What makes iridium rare in Earth rocks? Hint: it's a dense metal...
- grains of "shocked quartz" requires very high pressure
- spherical rock droplets how does that happen?
- soot from forest fires

#### Impacts and Mass Extinctions on Earth

- 1. debris in atmosphere blocks sunlight; plant die... animals starve
- 2. poisonous gases form in atmosphere
- 3. Other ancient 'mass extinctions' show an impact cause





Chicxulub impact - 65 M years ago

# When Did Humans Come Along?

- Life formed ~3.5 Billion yr ago
- Atmosphere changed ~2 bya)
- Diversity explosion (0.54 bya)
- Humans ~few million yr ago)

"Humans are but an insignificant twig on a vast arborescent bush, which, if planted again from seed, would unlikely reproduce us, or anything like us" -Steven Jay Gould





### Why does it feel so crowded all of a sudden?

Consider also 'the rate of change' of society, technology



# Life elsewhere in our solar system?



# Habitable Zones around stars



... the range in "distance from a star" at which liquid water can exist on the planet's surface for billions of years

# Odds seem good for planets within habitable zones...





• only one in Habitable Zone (almost?)





#### The Drake Equation Number of civilizations with which communication might be possible

The Drake equation states that:

$$N = R^* \times f_p \times n_e \times f_\ell \times f_i \times f_c \times L$$

where:

N = the number of civilizations in our galaxy with which communication might be possible;

and

R = the average rate of star formation per year in our galaxy

Ip = the fraction of those stars that have planets

no = the average number of planets that can potentially support life per star that has planets

 $I_{\ell}$  = the fraction of the above that actually go on to develop life at some point

II = the fraction of the above that actually go on to develop intelligent life

Ic = the fraction of civilizations that develop a technology that releases detectable signs of their existence into space

L = the length of time such civilizations release detectable signals into space.<sup>[3]</sup>

• Numbers we could put in this equation are very, very poorly known - basically, wild guesses

#### • Biggest unknown is probably how long civilizations last

http://www.youtube.com/watch?v=0Ztl8CG3Sys&feature=related Carl Sagan on Drake Eqn.

